

***Metrological Traceability  
&  
Measurement Uncertainty***

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**WIEN  
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*Comparability* of results of chemical amount measurements is needed in:

- determination of the value of goods in border-crossing trade
- implementation of border-crossing environmental regulations (water, air, soil)
- verification of safety of border-crossing food
- border-crossing use of clinical measurements
- application of forensic science across borders
- detection of border crossing adulterated products
- verification of authenticity of border-crossing products

Measurement results can only be *compared* if they have been obtained against a common "stated reference".



# What is traceability ?

# Key 1 for Metrology in Chemical Measurement:

## *Traceability*

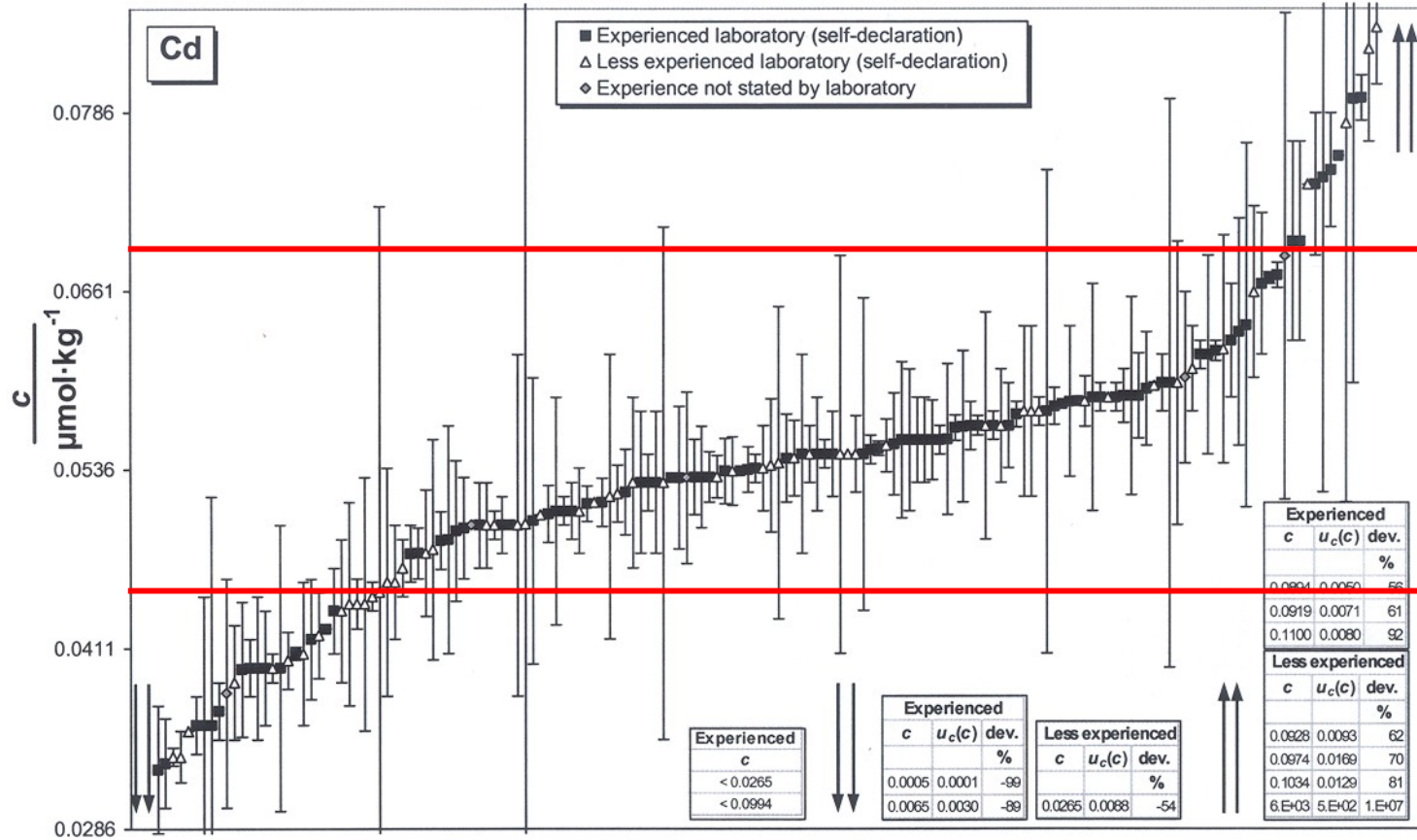
*property of the result of a measurement or the value of a standard whereby it can be related to stated references, usually national or international standards, through an unbroken chain of comparisons all having stated uncertainties*

VIM 1993

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The logo for the International Union of Pure and Applied Chemistry (IUPAC) and the International Union of Pure and Applied Metrology (IUM). It consists of the letters 'i', 'r', and 'm' in a stylized, lowercase, blue font. The 'i' and 'r' are positioned above the 'm', and the 'm' is positioned below the 'r'. The letters are interconnected and have a slight shadow effect.

IMEP- 6 : Trace elements in water : synthetic water (sample " 1")  
 Certified range ( $= \pm 2 u_c$ ) : 0.056 1 - 0.058 3  $\mu\text{mol}\cdot\text{kg}^{-1}$



172 results from all laboratories arranged by ascending values.



## Key 2 for Metrology in Chemical Measurement:

### Uncertainty

#### Type A evaluation:

method of evaluation of a standard uncertainty by the statistical analysis of a series of observations

#### Type B evaluation:

method of evaluation of a standard uncertainty by means other than the statistical analysis of a series of observations

ISO BIPM GUM 1993

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The logo consists of the letters 'i', 'r', and 'm' in a stylized, overlapping arrangement. The 'i' is on the left, the 'r' is in the middle, and the 'm' is on the right. The letters are in a dark blue color with a white outline.

PDB9843

# Current revision of the VIM:

## 1. improvement through decrease of

### - inconsistencies:

- internal within VIM
- internal within ISO Guides and Standards
- external between ISO- and non-ISO documents

### - unclarities, since: ...

- no clarity



- no understanding



- no agreement possible

*in trade*

*in implementation of EC Directives*

*in mutual acceptance of measurement results*



## ***The (revision of VIM-2, yielding) VIM-3***

2. takes on board chemical measurement
3. uses the “substitution principle”:  
any definition of a term for the definition of a concept must be capable of replacing that term in another definition
4. uses GUM, even if a number of GUM terms for concepts is already up for refinement in “GUM-2” (which is not yet planned)

## ***The (revision of VIM-2, yielding) VIM-3***

5. obviously still suffers from a carry-over from 20th - and even 19th - century thinking
6. contains “counting quantities is an implicit base quantity”
7. makes uncertainty part of the measurement result
8. defines “metrological traceability”
9. defines “traceability to the SI”
10. defines “measurand” as a “quantity intended for measurement
11. includes thinking in terms of “chemical measurement”  
(for the first time !)
12. includes examples of chemical measurements  
(for the first time)
13. introduces “calibration hierarchy”
14. gives a definition of “comparability”
15. specifies any “CRM” as a “measurement standard”

## Some examples justifying a revision:

example 1: “measurand”

example 2: “measurement result”

example 3: “metrological traceability”

example 4: “measurement uncertainty”

example 5: “measurement unit” and “measurement scale”

example 6: “metrological comparability”

example 7: “target measurement uncertainty”

## Key 2: Measurement uncertainty (MU)

Measure of doubt about the measurement result.

1. an uncertainty of a measurement result is generated automatically from the very moment one starts measuring
2. measurement uncertainty is obtained through an evaluation process:
  - **type A evaluation**
  - **type B evaluation**

(calculation comes only in at the end, **after** evaluation)

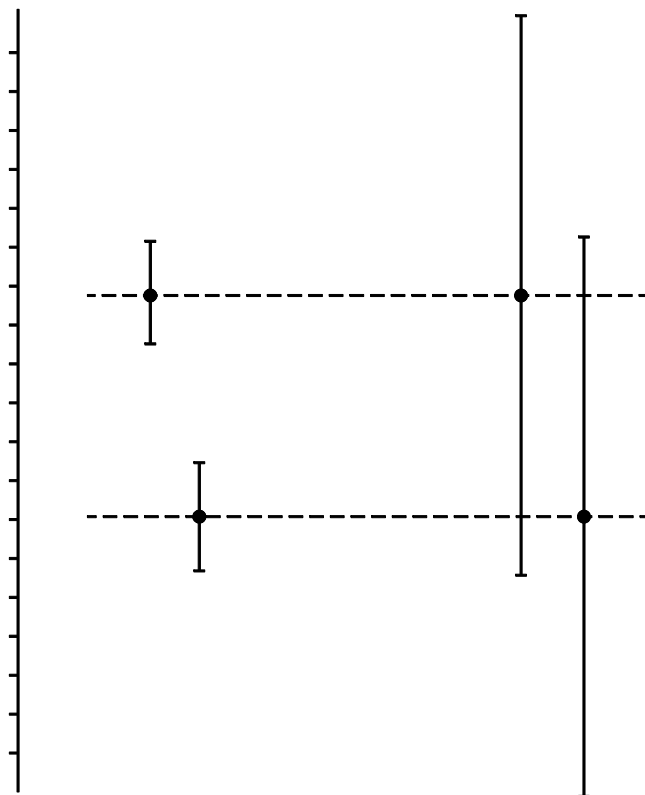
3. understanding the process called measurement, is a prerequisite for evaluation of measurement uncertainty
4. measurement uncertainty should be small enough for the intended use but, need not be smaller (it *can*, but *need not* be smaller)

# ***Uncertainty is new to chemists !***

- absent in most university curricula
- ISO/BIPM guide (GUM): only 12 years old !
- Repeatability is thought of as uncertainty
- Terra incognita !

# ISO/BIPM-guide: the new approach

re-establishes the responsibility  
of the analyst because  
“evaluation of uncertainty”  
is a thinking process !



**DISCREPANCY**

**NO DISCREPANCY**

**PROBLEM**

**NO PROBLEM**

**DISCREPANCY PROBLEMS CAN BE CAUSED  
UNNECESSARILY BY LACK OF  
"ORTHODOX" UNCERTAINTY ASSESSMENT**



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SPECIAL ISSUE PAPER

Paul De Bièvre · Philip D. P. Taylor

**“Demonstration” vs. “designation” of measurement competence:  
the need to link accreditation to metrology**

**Metrology**

**creates**

**respect for the law**

**Lack of Metrology**

**creates**

**contempt of the law**

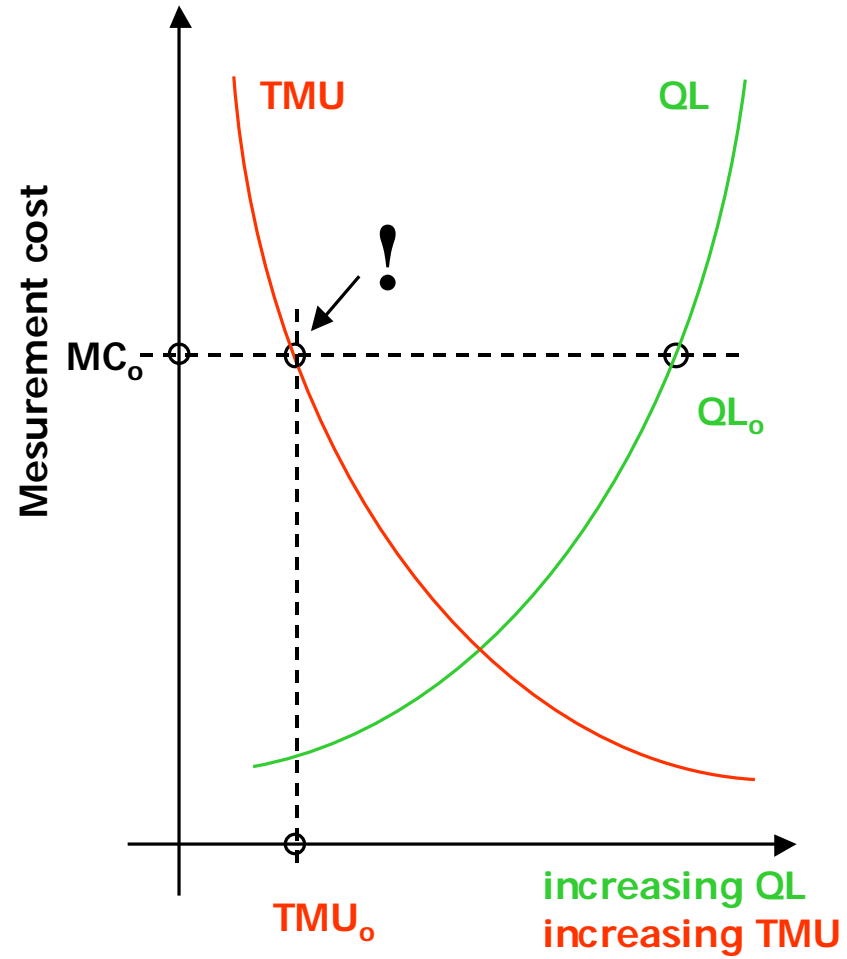
**The task is  
to demonstrate  
the authority of the result  
of the measurement.**

**This needs  
underpinning (“Untermauerung”)  
of the measurement result  
in order to lead to  
the necessary credibility.**

### Key 3: Target Measurement Uncertainty (TMU)

Range of uncertainties from which the analyst claims it contains a specified “traceable” value for the measurand

1. The analyst needs a “target” for the measurement uncertainty he must attempt to achieve
2. A TMU usually originates from a requirement put down by regulatory authorities

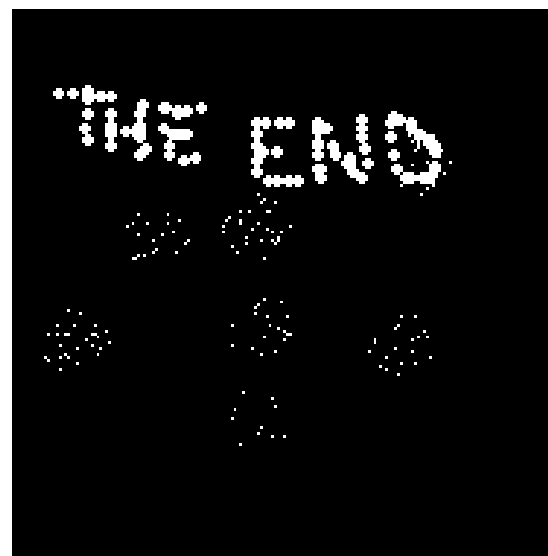


# Metrology in chemistry : The Hubble telescope ...

**BEFORE**



**AFTER**



= the 'spectacles' of the analytical chemist