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PURE AND APPLIED CHEMISTRY**

AND

**INTERNATIONAL FEDERATION OF  
CLINICAL CHEMISTRY**

IUPAC SECTION ON CLINICAL CHEMISTRY  
COMMISSION ON QUANTITIES AND UNITS  
AND

IFCC COMMITTEE ON STANDARDS  
EXPERT PANEL ON QUANTITIES AND UNITS

**LIST OF QUANTITIES IN  
CLINICAL CHEMISTRY  
RECOMMENDATION 1973**

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## PREFACE

The Commission on Quantities and Units in Clinical Chemistry\* is a part of the Section on Clinical Chemistry of the International Union of Pure and Applied Chemistry (IUPAC). The Expert Panel on Quantities and Units† is a part of the Committee on Standards of the International Federation of Clinical Chemistry (IFCC). These two bodies, the Commission and the Expert Panel, have worked jointly on this document—the former mainly concerned with basic philosophy, the latter with problems of implementation.

The aim of this document is to serve as a guide in supplanting present vernacular names for measurable properties in clinical chemistry. Systematic and more informative names are recommended, based on chemical and biochemical nomenclature. At the same time, a preference for 'molecular' kinds of quantities and SI units is stressed.

The tentative version of the present publication appeared as a IUPAC-IFCC (yellow) *Information Bulletin*, Appendices on Tentative Nomenclature, Symbols, Units, and Standards, No. 21, February 1972. As a consequence of comments from knowledgeable colleagues and reconsiderations by the Commission and Expert Panel a few minor changes have been made for this Recommendation 1973.

IUPAC has approved this Recommendation 1973. The IFCC Committee on Standards and Executive Board have also approved, but the document still needs the confirmation of the IFCC Council.

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## **ABBREVIATIONS OF REFERENCES**

CQUCC	Commission on Quantities and Units in Clinical Chemistry
EPQU	Expert Panel on Quantities and Units
ICSH	International Committee for Standardization in Hematology
IFCC	International Federation of Clinical Chemistry
INN	International Non-Proprietary Name
IUB	International Union of Biochemistry
IUPAC	International Union of Pure and Applied Chemistry
NFN	Det nordiske Farmakopénævn
QU-R66	Quantities and Units in Clinical Chemistry. Recommendation 1966 (IUPAC-IFCC) (8.2)
WAPS	World Association of (Anatomic and Clinical) Pathology Societies

## 0. INTRODUCTION

0.1.—In 1967 a monograph (8.2) (QU-R66) appeared containing a *Recommendation 1966* of the Commission on Clinical Chemistry of the International Union of Pure and Applied Chemistry (IUPAC) and of the International Federation of Clinical Chemistry (IFCC). Since then, the work has been continued by the Commission on Quantities and Units in Clinical Chemistry (CQUCC) of IUPAC and by the Expert Panel on Quantities and Units (EPQU) of IFCC (cf. footnotes on p. 549). A condensed and revised version of QU-R66 is printed as a separate Recommendation (8.5).

0.2.—Concerning names of clinical chemical quantities, the *Recommendation 1966* stated (QU-R66-6.3.1) that *the generic part of the quantity name be unambiguous and contain the following three segments of information:*

- kind of system* of which a component or function is measured, e.g.  
Serum
- component*, e.g. Sodium ion
- kind of quantity*, e.g. 'molar concentration', now called substance concentration (cf. 5.2.1).

This statement is now supported by a joint Recommendation 1972 from ICSH, IFCC, and WAPS (8.3).

0.3.—On the basis of a thorough discussion (QU-R66-2.3) it was also stated in the *Recommendation 1966* (QU-R66-4.6) that the kind of quantity *amount of substance* is of 'central importance in clinical chemistry'. Consequently, this basic kind of quantity (measured in mole) and derived kinds of quantities, e.g. substance concentration and mole fraction, are preferred for mass, mass concentration, mass fraction, respectively. This recommendation is now also joint ICSH, IFCC, and WAPS (8.3).

0.4.—These principles, supplemented by some suggested rules, were applied to compile a List of Quantities (QU-R66-7) where the complete systematic names of one hundred quantities were given as examples for discussion.

0.5.—The present new list of names is the outcome of further deliberations in CQUCC and EPQU and also incorporates various suggestions from other sources as well as developments in international nomenclature since 1966.

## 1. SCOPE OF LIST

The quantities listed are meant to cover many, but not all, of those measured today in clinical chemical laboratories. It is hoped that names for quantities omitted from the list may be created easily from analogous entries. Not every example is necessarily clinically relevant.

Methods of reporting loading test results, e.g. 'glucose tolerance', are not specifically given. They may contain information on the procedure followed and a table or graph of paired results for time and clinical chemical quantity.

As many clinical chemical laboratories are concerned also with determination of hematological, clinical physiological, clinical microbiological and toxicological quantities, some of these are included.

## 2. STRUCTURE OF LIST

2.1.—Each entry is given an alpha-numeric serial reference, e.g. B17, for convenience of identification in this document.

2.2.—The general form of an entry is according to *Recommendation 1966* (QU-R66-3.1-1)

$$\text{quantity} = \text{number} \times \text{unit}$$

2.3.—Most of the quantity names listed have the recommended information arranged in the following form

System—Component, kind of quantity

Specifications in parentheses may be necessary to one or more of the three segments. More complex types of names occur, e.g. [A02, A13].

2.4.—The quantities are listed alphabetically according to the component that, arbitrarily, has been printed in small capitals.

2.5.—The value for each quantity is meant to represent a result from a 'normal' adult person.

2.6.—A set of printing rules is given in *Recommendation 1966* (QU-R66-3.8). The special problems inherent in the use of typewriters with restricted sets of characters have not been treated.

## 3. SYSTEMS

3.1.—The system employed is supposed to be a certain 'normal' person or a part of same (e.g. Blood, Leukocytes, Bilirubins) or some material the person takes up or excretes (e.g. Food, Faeces). Occasionally, a measuring system is used [C13].

3.2.—The following codes are used in this document (for others, cf. 8.5).

*Table 3.2-1. Codes for description of system*

B	Blood (in vessels)
aB	Arterial blood (in arterial vessels; including 'capillary blood')
vB	Venous blood (in venous vessels)
d	24-hour (change of system during 24 hours)
fPt	Fasting patient (measurement in the morning)
F	Faeces
P	Plasma (blood plasma)
Pt	'Patient' (individual on whom measurement is being made)
S	Serum (blood serum)
Sf	Spinal fluid
U	Urine

Too many abbreviations and codes should be avoided and, in case of doubtful comprehensibility, the full name of the system should be given.

## LIST OF QUANTITIES IN CLINICAL CHEMISTRY

It may be necessary to indicate that the system of the quantity is part of a supersystem: (aB)P means that the plasma of the quantity is derived from arterial blood [D04].

When specifications are necessary, they are placed in parentheses after the system [D03, E04].

3.3.—The calendar time of obtaining the specimen should be a part of the identification of the person as well as other relevant information on technique of obtaining the specimen plus the physiological state of the person. That the specimen was obtained in the morning and that the person was fasting is symbolized fPt. Thus, (fPt)S or ((fPt)B)S means that the serum of the quantity is derived from blood of the fasting person in the morning [A17].

The time (interval) during which a registered change occurred in a system is coded in one situation: d for change during 24 hours = 1 d (morning to morning) [A18]. In other cases, the calendar time interval should be specified, e.g. after the kind of quantity.

## 4. COMPONENTS

4.1.—Names of components should be printed in full.

4.2.—The following types of components may be distinguished:

4.2.1.—*Definable chemical* components or *groups of such* are built of characteristic steric structures which may be symbolized by a chemical formula or common elementary entity, e.g. Carbamide, CO(NH<sub>2</sub>)<sub>2</sub>; Amino acid nitrogen, N.

4.2.2.—*Functional chemical* components are chemical compounds or groups of compounds characterized by common properties or effects, e.g. Alkaline phosphatase.

4.2.3.—*Physical* components are composed of micro- or macroscopic physical bodies, which constitute particles or phases in the system, e.g. Erythrocytes.

4.2.4.—Chemical or physiological processes may serve as components, e.g. Coagulation, Capillary bleeding [C34, C09].

4.3.1.—Chemical names are in accordance with IUPAC recommendations.

4.3.2.—Sodium is the name of the metallic element with atomic number 11 and oxidation number zero. This component is not found in biological systems, whereas *Sodium ion* is present.

4.3.3.—If doubt may arise as to oxidation number the *Stock notation* is employed using Roman numerals in parentheses. Thus, Calcium(II) signifies calcium with oxidation number + II irrespective of whether it be free ion Ca<sup>2+</sup> or different kinds of chelated calcium(II) with ionization charges more difficult to define [C04, C05, C06, C07].

4.3.4.—*Roman numerals* are sometimes used for other kinds of specifications to the component name, e.g. for porphyrins or coagulation factors [U11, C35].

4.3.5.—*Trivial chemical names* have often been chosen for brevity instead of systematic chemical names. Where several trivial names exist for a substance the more ‘chemical’, i.e. the more informative, is given, e.g. carbamide for urea [C10].

4.3.6.—*Prefixes defining stereoisomeric structure* are omitted if not essential, e.g. Glucose for D-Glucose.

4.4.—For *drugs*, the names from national or, preferably, international pharmacopoeas should be used (NFN, INN).

4.5.—*Enzymes* are given Recommended Names according to the *Recommendation 1972* from International Union of Biochemistry (8.1) [A09].

4.6.—*Blood cells* are named according to suggestions by the Working Group on Standardized Documentation of Hematological Findings of the Expert Panel on Hematological Documentation under the International Committee for Standardization in Hematology (ICSH) [B08, E01, P08]. This nomenclature is not yet accepted by ICSH, but seems short, logical, and directly understandable.

4.7.—Taxonomic names for *families* and *genera* are printed in italics [M16].

4.8.—For *acids and bases*, defined according to Brønsted, the totality of corresponding acid–base pairs or series are often considered as one group. No accepted rules exist for naming such mixtures. The following rules are recommended:

4.8.1.—The designation of the maximally ionized form of those in question is given, omitting the word ‘ion’. Thus, Ammonium [A25] comprises ammonium ion and ammonia; Creatininium [C47] comprises creatininium ion and creatinine; Ascorbate [A33] comprises ascorbate ion and ascorbic acid; Carbonate [C18] comprises carbonate ion, hydrogen carbonate ion, and carbonic acid, but *not* carbon dioxide. It should be noted that, in contrast to Ammonia and Ascorbic acid, the designations Ammonium and Ascorbate do not indicate single well-defined, chemical compounds.

4.8.2.—Trivial names for organic amphi-ionic substances having trivial names are used for the totality of the amphi-ionic, acid, and base forms. E.g. Hydroxyproline [H22] means hydroxyprolinium ion plus Hydroxyproline (non-charged) plus Hydroxyprolinate ion.

4.8.3.—Mixtures of a defined chemical component and its derivatives may be denoted by the plural form of the name of the pure substance [B13, C30].

4.9.—The bare component name may have to be further *specified* in parentheses to avoid ambiguity.

The elementary entity [C37, P02] or relative molecular mass [C05, T06] is often necessary information. The latter alternative is especially useful for proteins where the structure is not known in detail, and where the use of mass concepts also needs specification. For mixtures of substances limits may be specified [A17, A18, C35, U11]. To indicate the sum of components, specified in individual quantities, the specification ‘total’ may be applied [C08]. Also, specifications relating to a physiological process in which the

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substance participates may be needed [D05].

4.10.—In many cases, several equally correct names are possible for the same component [A21, N08] [C04, C08]. Which one is preferred depends on the context.

### 5. KINDS OF QUANTITIES

5.1.—The kinds of quantities used are those given in *Recommendation 1966* with amendments and additions based on recent IUPAC and IUB recommendations, and approved by CQUCC and EPQU (cf. 8.5).

5.2.1.—The English designation ‘molar’ is now restricted by IUPAC to kinds of quantities with a definition containing the expression ‘an extensive kind of quantity of the system divided by amount of substance of the system’. Consequently, ‘molar concentration’ is no longer acceptable; the systematic name is a cumbersome ‘amount of substance concentration’ or ‘concentration’ (8.4). However, after consultation with IUPAC Commission on Symbols, Terminology, and Units it was agreed that CQUCC can recommend the name ‘substance concentration’ to avoid confusion with the word ‘concentration’ in the colloquial sense.

5.2.2.—The IUPAC-IUB Recommendations 1972 (8.1) contain recent changes in some of the names and definitions for kinds of quantities and units in enzymology. They define the derived kind of quantity ‘enzymic activity’ as ‘the rate of reaction of substrate that may be attributed to catalysis by an enzyme’ and have the derived unit ‘katal’ = 1 mol/s. This approach creates dimensional difficulties (8.5) and, after much discussion, CQUCC and EPQU have decided to retain the former suggestion of ‘catalytic amount’ considered as a basic kind of quantity with a base unit ‘katal’, symbolized kat and defined as the catalytic amount of any catalyst (including any enzyme) that catalyses a reaction rate of one mole per second in an assay system. It should be stressed that numerical values are identical whether one uses the basic kind of quantity ‘catalytic amount’ and base unit ‘katal’ or the derived kind of quantity ‘enzymic activity’ and derived unit ‘katal’ = 1 mol/s. The use of the ‘enzyme unit’ (symbolized U)  $\triangleq$  1  $\mu$ mol/min should be progressively discouraged. For a given method 1 U  $\triangleq$  16,67 nkat. The derived kind of quantity ‘catalytic concentration’ (CBN: concentration of enzymic activity) uses the non-coherent unit kat/l; 1 U/ml  $\triangleq$  16,67  $\mu$ kat/l.

5.2.3.—CQUCC in Washington, DC, 1971, decided that the kind of quantity ‘number of particles’ (QU-R66-4.5) be renamed ‘number (of entities)’. In accordance, ‘particle concentration’ (QU-R66-4.14) and the ambiguous ‘particle fraction’ (QU-R66-4.15) were renamed ‘number concentration’ and ‘number fraction’ respectively.

5.3.—A few recommended kinds of quantities that have not been systematically described by CQUCC are used in the List with recent modifications.

5.3.1.—*Volume content* (QU-R66-5.1. Specific volume content) is the volume of the isolated component at specified conditions divided by the mass of the system [B16].

5.3.2.—*Mean substance rate* (QU-R66-5.5. Mole rate) is the amount of substance of the component changed in or moved to or from a system divided by the time during which the component was changed or moved [cf. D05].

5.3.3.—*Mean volume rate* (QU-R66-5.4. Volume rate) is defined analogously with 5.3.2, substituting ‘volume’ for ‘amount of substance’ [F03].

5.4.1.—For so-called *qualitative tests*, a quantitative expression is used, employing a kind of quantity of an arbitrary nature (QU-R66-4.26). The two possible numerical values are given as specification to the kind of quantity: (0–1) [A27]. The symbols + and – are not recommended. (Cf. also 5.7.1 and 5.7.3.)

5.4.2.—Another, equally acceptable possibility is the use of, e.g., ‘substance concentration’ without specifications and a result showing whether the quantity is larger than or smaller than the substance concentration indicating the methodological limit of detection, e.g. Urine—Glucose, substance concentration < 12 mmoll instead of 0.

5.5.—So-called *semiquantitative tests* are quantitative and should be treated as such [B14], cf. also 5.7.3.

5.6.1.—In the List, all names of kind of quantity are given *unabbreviated*. If it be necessary to save space, the following three possibilities may be considered.

5.6.2.—The name of the kind of quantity may be *omitted with reference to a local rule, provided that misunderstandings are impossible in practice*. E.g. Plasma—Glucose = 8,2 mmol/l would be permissible if the data sheet states that: (i) the kind of quantity employed is given in the local laboratory manual or (ii) all results in ‘mmol/l’ refer to the kind of quantity ‘substance concentration’.

5.6.3.—The kind of quantity may be given as its recommended symbol (cf. QU-R66-4). E.g. Plasma—Glucose,  $c = 8,2 \text{ mmol/l}$ . All these symbols are in italics (underlined in typewritten text). The meaning of the symbols may be printed on the data sheet or in the laboratory manual.

5.6.4.—The name for the kind of quantity is suitably abbreviated, cf. QU-R66-7.3-2, substituting ‘substc.’ for ‘mole.’ as abbreviation for ‘(amount of) substance concentration’.

5.7.1.—*Specifications* are given in parentheses. The specific form shown is short, but should not be taken as a recommendation.

5.7.2.—For some kinds of quantities, the method used is a necessary specification: (i) kinds of quantities of an arbitrary nature; (ii) kinds of quantities relating to a component characterized by its function, e.g. an enzyme.

In other cases the inclusion or omission of the method is a matter of judgement about the dependence of the results on the current method(s).

5.7.3.—If results are restricted to a scale of integers, the closed interval of possible results should be stated, cf. 5.4.1 and 5.5 [A27, B14].

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5.7.4.—In the case of quantities involving ‘concentration’ in liquids, the temperature of the system has been omitted because it is considered not to be sufficiently important.

5.7.5.—In practice, specifications should be adapted to local needs. A number referring to the laboratory manual, which contains all relevant information, is the shortest specification possible.

## 6. NUMBERS

6.1.—The numbers given in the List (value of quantity divided by unit) serve as illustrations and, presumably, contain only significant figures in accordance with the rule that the analytical disturbance should influence only the last figure.

6.2.—For some ‘arbitrary’ quantities, the result is always an integer, including ‘zero’, cf. 5.4.1 and 5.5 [A27].

6.3.—Where method is decisive for the magnitude of the result, the number is supplanted by a question mark, e.g. in the case of enzymes [A02].

## 7. UNITS

7.1.—In each case, the size of the unit has been chosen in accordance with the biological level and the estimated analytical variation of the result. Only factors that are  $10^{3n}$ , where  $n$  is an integer, should be used.

The units kat, kat/l, and kat/mol are used for enzymes throughout for the reason stated in 6.3, but in most cases sub-units will be required.

7.2.—Quantities having the dimension one use the base unit ‘unity’ (1) and one of the units  $10^{3n}$  is given, e.g. 1 or  $10^{-3}$  [A07, B06].

7.3.—As a consequence of IUPAC recommendations (8.4) the following deviations from QU-R66 have been made:

7.3.1.—‘Thermodynamic temperature’ and ‘temperature difference’ are given in the base unit kelvin (K), not in degree Kelvin ( $^{\circ}\text{K}$ ) [R01, F08].

7.3.2.—‘Pressure’ and ‘partial pressure’ use the unit pascal (Pa), which is identical with ‘newton per square metre’ ( $\text{N/m}^2$ ) [B17].

## 8. REFERENCES

(Abbreviations, see p. 551)

- 8.1. Commission on Biochemical Nomenclature of the IUPAC and the IUB, *Enzyme Nomenclature. Recommendations (1972) on the Nomenclature and Classification of Enzymes together with their Units and the Symbols of Enzyme Kinetics*, 443 pp, Elsevier: Amsterdam (1973).
- 8.2. R. Dybkær and K. Jørgensen, *Quantities and Units in Clinical Chemistry Including Recommendation 1966 of the Commission on Clinical Chemistry of the IUPAC and of the IFCC*, x + 102 pp, Munksgaard: Copenhagen (1967).

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- 8.3. International Committee for Standardization in Hematology, International Federation of Clinical Chemistry and World Association of (Anatomic and Clinical) Pathology Societies: Recommendation for Use of SI in Clinical Laboratory Measurements (1972). (See, e.g., *Z. klin. Chem.*, **11**, 93 (1973) (only).)
- 8.4. International Union of Pure and Applied Chemistry, Division of Physical Chemistry, Commission on Symbols, Terminology, and Units, *Manual of Symbols and Terminology for Physicochemical Quantities and Units*, 44 pp, Butterworths: London (1970). (Also in: *Pure and Appl. Chem.*, **21**, 1-44 (1970).)
- 8.5. International Union of Pure and Applied Chemistry, Section on Clinical Chemistry, Commission on Quantities and Units and International Federation of Clinical Chemistry, Committee on Standards, Expert Panel on Quantities and Units, *Quantities and Units in Clinical Chemistry. Recommendation 1973*, Butterworths: London, to be published. (Also in: *Pure and Appl. Chem.*, **37**, 517 (1974).)

Component names in Section 9 are given in small capitals, e.g. ACETO-ACETATE. This does not constitute a recommendation. It is often preferable to use ordinary type or (boldface) heavy type, but with an initial capital, e.g. Acetoacetate or **Acetoacetate**.

The letter l and numeral 1 should have distinctly different forms in type-written and printed texts. Unfortunately, this is not possible in the fount used for the present document.

*Note:* The choice between spelling with -æ- or -œ- or -e- is not meant to imply a preference.

## 9. ALPHABETICAL LIST OF QUANTITIES (Synonyms are found in Section 10)

Ser. No.	System—Component(specifications), kind of quantity(specifications) =	Number × Unit	Ref.
A01	S—ACETOACETATE, substance concentration	22 $\mu\text{mol/l}$	
A02	(B)Erythrocytes—ACETYLCHOLINESTERASE, catalytic amount <i>per</i> HEMOGLOBIN(Fe), amount of substance	? kat/mol	
A03	S—ACETYLCHOLINESTERASE, catalytic concentration(method)	? kat/l	
A04	dU—ACID(H), amount of substance(Jørgensen 1957)	40 mmol	
A05	S—ACID PHOSPHATASE, catalytic concentration(method)	? kat/l	
A06	S—ACID PHOSPHATASE(TARTRATE SENSITIVE), catalytic concentration(method)	? kat/l	
A07	(dU)Adrenalinium + noradrenalinium	1 $\mu\text{mol}$	
	—ADRENALINUM, mole fraction	0,2 $\mu\text{mol}$	
A08	dU—ADRENALINUM + NORADRENALINUM, amount of substance	0,30 kat/l	
A09	S—ALANINE AMINOTRANSFERASE, catalytic concentration(method)	? g/l	A14
A10	S—ALBUMIN, mass concentration	42 g/l	A15
A11	Sf—ALBUMIN, mass concentration	0,20 g/l	
A12	(S)Protein—ALBUMIN, mass fraction(method)	0,60 1	
A13	S—ALBUMIN, mass per GLOBULIN, mass	1,5 1	
A14	S—ALBUMIN(68 000), substance concentration	618 $\mu\text{mol/l}$	A10
A15	Sf—ALBUMIN(68 000), substance concentration	2,9 $\mu\text{mol/l}$	A11
A16	dPt—ALDOSTERONE(PRODUCED), amount of substance(method)	138 nmol	
A17	(fPt)S—ALIPHATIC CARBOXYLATE( $C_{10}-C_{26}$ ), NON-ESTERIFIED, substance concentration	0,58 mmol/l	L11
A18	dF—ALIPHATIC CARBOXYLATE( $C_{10}-C_{26}$ ), NON-ESTERIFIED + ESTERS, amount of substance	15 mmol/l	
A19	(fPt)S—ALIPHATIC CARBOXYLATE( $C_{10}-C_{26}$ ), NON-ESTERIFIED + ESTERS, substance concentration	11,1 kat/l	
A20	S—ALKALINE PHOSPHATASE, catalytic concentration(method)	? N08	
A21	(fPt)S—AMINO ACID-NITROGEN, substance concentration	3,0 mmol/l	
A22	dU— $\delta$ -AMINOEVULINATE, amount of substance	43 $\mu\text{mol}$	Note 1
A23	S— $\delta$ -AMINOEVULINATE, substance concentration	1,2 $\mu\text{mol/l}$	
A24	S—AMINOPEPTIDASE(CYTOSOL), catalytic concentration(method)	? kat/l	Note 2
A25	dU—AMMONIUM, amount of substance	51 mmol	
A26	P—AMMONIUM ion, substance concentration	30 $\mu\text{mol/l}$	
A27	Rectal mucus—AMOeba, arbitrary number concentration(method; 0-1)	0 arb. unit	
A28	U— $\alpha$ -AMYLASE, arbitrary catalytic concentration(method)	? arb. unit	
A29	S— $\alpha$ -AMYLASE, catalytic concentration(method)	? kat/l	
A30	S—ANION( $\text{A}^-$ ), substance concentration	155 mmol/l	
A31	S—ANTITHROMBIN III, arbitrary substance concentration(method)	? arb. unit	
A32	S— $\alpha_1$ -ANTITRYPsin, arbitrary substance concentration(method)	? arb. unit	
A33	(fPt)S—ASCORBATE, substance concentration	55 $\mu\text{mol/l}$	
A34	S—ASPARTATE AMINOTRANSFERASE, catalytic concentration(method)	? kat/l	

Note 1: The component is also called 4-Oxo-5-aminopropanoate.

Note 2: EC 3.4.11.1; the component was formerly called leucine aminopeptidase.

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Ser. No.	System	Component(specifications), kind of quantity(specifications)	Number	$\times$	Unit	Ref.
B01	U—BACTERIA, arbitrary number concentration(method; 0-1)	0		arb. unit		
B02	U—BACTERIA(LIVING), number concentration(method)	10		$10^6/l$		
B03	S—BARBITURATES, arbitrary substance concentration(method)	0		arb. unit		Note 3
B04	aB—BASE(H <sup>-</sup> -BINDING GROUPS), substance concentration(Singer and Hastings 1948)	45		mmol/l		
B05	aB—BASE(H <sup>+</sup> -BINDING GROUPS), substance concentration difference(method; Pt - Norm)	0,0		mmol/l		
B06	Erythrocytes—BASOPHIL PUNCTATED ERYTHROCYTES, number fraction	0,20		$10^{-3}$		
B07	B—BASOPHILOCYTES, number concentration	0,12		$10^9/l$		
B08	(B)Leukocytes—BASOPHILOCYTES, number fraction	0,01		1		
B09	S—BILIRUBIN, substance concentration	4		$\mu\text{mol}/l$		
B10	(S)Bilirubins—BILIRUBIN ESTER, mole fraction(method)	0,5		1		
B11	S—BILIRUBIN ESTER, substance concentration	5		$\mu\text{mol}/l$		
B12	U—BILIRUBINS, arbitrary substance concentration(method; 0-1)	0		arb. unit		
B13	S—BILIRUBINS(NON-ESTERIFIED + ESTERS), substance concentration	9		$\mu\text{mol}/l$		
B14	F—BLOOD, arbitrary volume fraction(method; 0-4)	0		arb. unit		H06 Note 4
B15	Pt—BLOOD, volume	4,90		1		
B16	Pt—BLOOD, volume content(method)	70		ml/kg		
B17	Pt—BLOOD(ARTERIAL, DIASTOLIC), pressure(method)	10,4		kPa		
B18	Pt—BLOOD(ARTERIAL, SYSTOLIC), pressure(method)	15,6		kPa		
B19	B—BLOOD COAGULATION LYSIS, time(method)	?		s		
B20	S—BROMIDE, substance concentration	0,0		mmol/l		

Note 3: The numerical values equal those of 'S—Barbiturates, substance concentration', but measurement is performed as if the barbiturate(s) present were allylpropyl mal(NFN)—hence 'arbitrary'.

Note 4: The 'concentration' of blood is sought even if its hemoglobin is used as a marker in the method.

Ser. No.	System	Component(specifications), kind of quantity(specifications) =	Number	$\times$	Unit	Ref.
C01	dF—CALCIUM(II) (ca), amount of substance	16		mmol		
C02	dU—CALCIUM(II) (ca), amount of substance	5,2		mmol		
C03	dU,U—CALCIUM(II) (ca), substance concentration	4,3		mmol/l		
C04	S—CALCIUM(II) (ca), substance concentration	2,5		mmol/l		C08

**LIST OF QUANTITIES IN CLINICAL CHEMISTRY**

C05	S—CALCIUM(II) (Ca, CHELATE, < 1000), substance concentration	mmol/l
C06	S—CALCIUM(II) (Ca, NON-CHELATE), substance concentration	mmol/l
C07	S—CALCIUM(II) (Ca, PROTEIN BOUND), substance concentration	mmol/l
C08	S—CALCIUM(II) (Ca, TOTAL), substance concentration	mmol/l
C09	Pt—CAPILLARY BLEEDING, time(method)	ks
C10	dU—CARBAMIDE, amount of substance	mol
C11	dU—CARBAMIDE, substance concentration	mol/l
C12	S—CARBAMIDE, substance concentration	mmol/l
C13	Gas(aB equilibrated)—CARBON DIOXIDE, partial pressure(method; 37,0 °C)	mmol/l
C14	(aBP)—CARBON DIOXIDE, substance concentration	kPa
C15	Expiratory gas—CARBON DIOXIDE, volume fraction	mmol/l
C16	(B)Hemoglobin—CARBON MONOXIDE HEMOGLOBIN, mole fraction	1
C17	B—CARBON MONOXIDE HEMOGLOBIN(FE), substance concentration	mmol/l
C18	P—CARBONATE + CARBON DIOXIDE, substance concentration	mmol/l
C19	U—CASTS, arbitrary number, concentration(method; 0-1)	arb. unit
C20	U—CASTS, number(Addis 1949; 12 h, night)	10 <sup>3</sup>
C21	F—CATALASE, arbitrary catalytic concentration(method; 0-1)	arb. unit
C22	S—CATION(B <sup>+</sup> ), substance concentration	mmol/l
C23	SF—CELLS, number concentration	10 <sup>6</sup> /l
C24	S—CERULOPLASMIN(150 000), substance concentration	μmol/l
C25	dU—CHLORIDE, amount of substance	mmol
C26	dU,U—CHLORIDE, substance concentration	mmol/l
C27	S—CHLORIDE, substance concentration	mmol/l
C28	S—CHOLESTEROL, substance concentration	mmol/l
C29	S—CHOLESTEROL(NON-ESTERIFIED), substance concentration	mmol/l
C30	S—CHOLESTEROLES, substance concentration	mmol/l
C31	S—CHOLESTERYL ESTERS, substance concentration	mmol/l
C32	nU—CHORIONGONADOTROPIN, arbitrary substance concentration	pmol/l
C33	S—CITRATE, substance-concentration	arb. unit
C34	B—COAGULATION, time(method)	ks
C35	P—COAGULATION FACTOR(I + II + V + VII + X), arbitrary time(Quick 1935)	1
C36	P—COAGULATION FACTORS(I + II + VII + X), relative arbitrary substance concentration	1,00
C37	S—COBALAMIN(Co), substance concentration(method)	pmol/l
C38	S—COPPER(II), substance concentration	μmol/l
C39	dU—COPROPORPHYRINS(I + III), amount of substance	μmol
C40	P—CORTICOSTERONE, substance concentration	nmol/l

Note 5: This type of quantity equals C28, but the former may be used together with C31 for emphasis.

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Ref.	Number × Unit	System—Component(specifications), kind of quantity(specifications)																																																
C41	400 nmol/l	P—CORTISOL, substance concentration																																																
C42	32 nmol/l	P—CORTISONE, substance concentration																																																
C43	80 μmol/l	dU—CREATINE, amount of substance																																																
C44	67 μmol/l	dU,U—CREATINE, substance concentration																																																
C45	40 μmol/l	S—CREATINE, substance concentration																																																
C46	? kat/l	S—CREATINE KINASE, catalytic concentration(method)																																																
C47	19,5 mmol/l	dU—CREATININUM, amount of substance																																																
C48	16,3 mmol/l	dU,U—CREATININUM, substance concentration																																																
C49	0,07 mmol/l	S—CREATININUM, substance concentration																																																
C50	0 arb. unit	S—CRYOGLOBULIN, arbitrary substance concentration(method; 0-1)																																																
D01	8 nmol/l	P—1,4-DIOXYCORTISOL, substance concentration																																																
D02	0,0 mmol/l	dU—2,5-DIHYDROXYPHENYLACETATE, amount of substance																																																
D03	13,0 kPa	Gas(aB equilibrated)—DIOXYGEN, partial pressure(method; 37,0 °C)	D04	0,13 mmol/l	(aBP)—DIOXYGEN, substance concentration	D03	1																																											
D05	1,00 10 <sup>9</sup> /l	(P <sub>1</sub> )—DIOXYGEN(ABSORBED), relative substance rate (resting; Pt/Norm)	E01	0,22 1	B—EOSINOPHILOCYTES, number concentration	E02	0,02 1	(B)Leukocytes—EOSINOPHILOCYTES, number fraction	E03	1 arb. unit	U—EPITHELIAL CELLS, arbitrary number concentration(method; 0-3)	E04	7,5 μm	(B)Erythrocyte(means) (B)Erythrocyte(means)	E05	88 fl	—ERYTHROCYTE, diameter —ERYTHROCYTE, volume	E06	115 d	—ERYTHROCYTE LIFE, time	E07	1 arb. unit	U—ERYTHROCYTES, arbitrary number concentration(method; 0-3)	E08	0,7 10 <sup>6</sup> /l	nU—ERYTHROCYTES, number(Addis 1949; 12 h, night)	E09	5,00 10 <sup>12</sup> /l	B—ERYTHROCYTES, number concentration	E10	0 10 <sup>6</sup> /l	Sf—ERYTHROCYTES, number fraction	E11	0,47 1	B—ERYTHROCYTES, volume fraction	E12	0,035 μmol	dU—ESTRADIOL, amount of substance(female Pt, ovulation peak)	E13	0,082 μmol	dU—ESTRADIOL, amount of substance(female Pt, ovulation peak)	E14	0,030 μmol	dU—ESTROGEN, amount of substance(postmenopausal Pt)	E15	0,019 μmol	dU—ESTRONE, amount of substance(male Pt)	E16	0,0 mmol/l	B—ETHANOL, substance concentration

LIST OF QUANTITIES IN CLINICAL CHEMISTRY

Ser. No.	System	Component(specifications), kind of quantity(specifications) ...	Number × Unit	Ref.
F01	P—FIBRINOGEN(340 000)	substance concentration	8,8 μmol/l	
F02	P—FIBRINOLYSIS, time(method)	?	s	
F03	Glomeruli—FLUID(FILTRATED), mean volume rate(creatininium; 1d)	1,6 ml/s		
F04	Tap water—FLUORIDE, substance concentration	40 μmol/l		
F05	(B)Hemoglobin—FOETAL HEMOGLOBIN, mole fraction	0,00 1		
F06	(fPt)S—FOLATES, substance concentration(method)	26 nmol/l		
F07	U—FORMIMINOGLUTAMATE, amount of substance (after L-histidine, amount of substance = 97 mmol; 32,4 ks)	100 μmol/l		
F08	S—FREEZING-POINT DEPRESSION, temperature difference	54,5 mK		
F09	S—FRUCTOSE-BIOPHOSPHATE ALDOLASE, catalytic concentration(method)	?	kat	
G01	(B)Erythrocyte(mean)	?	kat	
G02	—GALACTOSE-1-PHOSPHATE URIDYL TRANSFERASE, catalytic amount(method)	2,0 arb. unit	arb. unit	Note 6
G03	S— $\alpha_1$ -GLOBULIN, arbitrary mass concentration	5,0 arb. unit	arb. unit	Note 6
G04	S— $\alpha_2$ -GLOBULIN, arbitrary mass concentration	3,0 arb. unit	arb. unit	Note 6
G05	S— $\beta_1$ -GLOBULIN, arbitrary mass concentration	3,0 arb. unit	arb. unit	Note 6
G06	S— $\beta_2$ -GLOBULIN, arbitrary mass concentration	8,5 arb. unit	arb. unit	Note 6
G07	S— $\gamma$ -GLOBULIN, arbitrary mass concentration	0 g/l	g/l	
G08	Sf—GLOBULIN, arbitrary substance concentration(method: 0-1)	22 mmol	mmol	
G09	dU—GLOBULIN, mass concentration	2,3 arb. unit	arb. unit	
G10	dU—GLUCOSE, arbitrary substance concentration(Clinistix $\oplus$ ; 0-1)	0 mmol/l	mmol/l	
G11	(fPt)P—GLUCOSE, substance concentration	5,6 mmol/l	mmol/l	
G12	Sf—GLUCOSE; substance concentration	2,8 mmol/l	mmol/l	
G13	U—GLUCOSE, substance concentration	1,9 mmol/l	mmol/l	
G14	(B)Erythrocyte(mean)—GLUCOSE-6-PHOSPHATE DEHYDROGENASE, catalytic amount(method)	?	kat	

Note 6: The mass concentration is not well known.

H01	S— $\alpha_2$ -HAPTOGLOBIN, arbitrary substance concentration(method: 0-3)	2	arb. unit
H02	S—HAPTOGLOBIN, substance concentration(hemoglobin binding)	7,5 μmol/l	
H03	Pt—height	1,70 m	Note 7

Note 7: An alternative is 'Pt—PATIENT, height' or 'Pt—BODY, height'.

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Ser. No.	System—Component(specifications), kind of quantity(specifications) =	Number × Unit	Ref.
H04	F— <i>H. heminthes</i> EGGS, arbitrary number concentration(method; 0-3)	0	arb. unit
H05	(B)Hemoglobin—HEMIGLOBIN, mole fraction	1	1
H06	F—HEMOGLOBIN, arbitrary substance concentration(method; 0-1)	0	arb. unit
H07	U—HEMOGLOBIN, arbitrary substance concentration(method; 0-1)	0	arb. unit
H08	(B)Erythrocyte(mean)—HEMOGLOBIN(Fe), amount of substance	2,0	fmol
H09	(B)Erythrocyte(mean)—HEMOGLOBIN(Fe), substance concentration	21	mmol/l
H10	P—HEMOGLOBIN(Fe), substance concentration	10	μmol/l
H11	B—HEMOGLOBIN(Fe), substance concentration(ICSH 1966)	9,6	mmol/l
H12	S—β <sub>2</sub> -HEMOPEXIN, arbitrary substance concentration(method; 0-3)	2	arb. unit
H13	U—HOMOGENITATE, arbitrary substance concentration(method; 0-1)	0	arb. unit
H14	P—HYDROGEN CARBONATE ION, substance concentration(blood; c(O <sub>2</sub> ) = 0,21 mmol/l; c(CO <sub>2</sub> ) = 1,19 mmol/l; θ = 37 °C)	23,6 39,6 10	mmol/l mmol/l μmol/l
H15	(aBP)—HYDROGEN ION, substance concentration	?	P06
H16	P—3-HYDROXYBUTYRATE, substance concentration	?	kat/l
H17	S—2-HYDROXYBUTYRATE DEHYDROGENASE, catalytic concentration(method)	0,42	mmol
H18	dU—17-HYDROXYCORTICOSTEROID, amount of substance(method)	0,42	μmol
H19	P—11-HYDROXYCORTICOSTEROID, substance concentration	30	μmol
H20	dU—5-HYDROXYINDOLYL ACETATE, amount of substance	25	μmol
H21	dU—4-HYDROXY-3-METHOXYINDOLE, amount of substance	0,16	mmol
H <sup>2+2-</sup>	dU—HYDROXYPROLINE, amount of substance	105	10 <sup>3</sup> int. unit/l
101	S—IMMUNOGLOBULIN A, arbitrary substance concentration(method)	120	10 <sup>3</sup> int. unit/l
102	S—IMMUNOGLOBULIN G, arbitrary substance concentration(method)	97	10 <sup>3</sup> int. unit/l
103	S—IMMUNOGLOBULIN M, arbitrary substance concentration(method)	25	10 <sup>-3</sup> int. unit/l
104	S—INSULIN, arbitrary substance concentration(method)	183	pmol/l
105	S—INSULIN(5800), substance concentration(method)	0,41	μmol/l
106	S—IODINE(I <sub>2</sub> PROTEIN BOUND), substance concentration	310	mmol/l
107	S—ION(A <sup>-</sup> + B <sup>+</sup> ), substance concentration	2,3	μmol
108	dU—IRON(II + III), amount of substance	1,9	μmol/l
109	dU, U—IRON(II - III), substance concentration		

LIST OF QUANTITIES IN CLINICAL CHEMISTRY

I10	$(\text{fP})S - \text{IRON}(\text{II} + \text{III})$ (IN HEMOGLOBIN AND TRANSFERRIN), substance concentration	33	$\mu\text{mol/l}$
I11	$(\text{fP})S - \text{IRON}(\text{III})$ (TRANSFERRIN BOUND), substance concentration	23,6	$\mu\text{mol/l}$
L01	S—LACTATE, substance concentration	0,50	$\text{mmol/l}$
L02	S—LACTATE DEHYDROGENASE, catalytic concentration(method)	?	kat/l
L03	dU—LEAD(II), amount of substance	0,07	$\mu\text{mol}$
L04	B—LEAD(II), substance concentration	1,2	$\mu\text{mol/l}$
L05	dU,U—LEAD(II), substance concentration	0,06	$\mu\text{mol/l}$
L06	U—LEUKOCYTES, arbitrary number concentration(method; 0-3)	1	arb. unit
L07	B—LEUKOCYTES, number concentration	6,5	$10^9/\text{l}$
L08	Sf—LEUKOCYTES, number concentration	1	$10^6/\text{l}$
L09	U—LEUKOCYTES + EPITHELIAL CELLS, number(Addis 1949; 12 h, night)	2,0	$10^6$
L10	F—LIPID, arbitrary volume fraction(method; 0-1)	0	arb. unit
L11	dF—LIPID, mass(method)	4,0	g
L12	(fP)S—LIPID, mass concentration(method)	6,7	g/l
L13	S— $\alpha_1$ -LIPOPROTEIN, arbitrary substance concentration(method; 0-3)	2	arb. unit
L14	S— $\alpha_2$ -LIPOPROTEIN, arbitrary substance concentration(method; 0-3)	2	arb. unit
L15	S— $\beta$ -LIPOPROTEIN, arbitrary substance concentration(method; 0-3)	2	arb. unit
L16	B—LYMPHOCYTES, number concentration	2,0	$10^9/\text{l}$
L17	(B)Leukocytes—LYMPHOCYTES, number fraction	0,30	1
M01	S— $\alpha_2$ -MACROGLOBULIN, arbitrary substance concentration(method; 0-3)	2	arb. unit
M02	dF—MAGNESIUM(II) (mg), amount of substance	7,2	mmol
M03	dU—MAGNESIUM(II) (mg), amount of substance	2,6	mmol
M04	dU,U—MAGNESIUM(II) (mg), substance concentration	2,2	$\text{mmol/l}$
M05	S—MAGNESIUM(II) (mg), substance conc	1,0	$\text{mmol/l}$
M06	S—MAGNESIUM(II) (mg, CHELATE, < 1000), substance concentration	0,13	$\text{mmol/l}$
M07	S—MAGNESIUM(II) (mg, PROTEIN BOUND), substance concentration	0,30	$\text{mmol/l}$
M08	S—MAGNESIUM(II) (mg, TOTAL), substance concentration	1,0	$\text{mmol/l}$
M09	S—MAGNESIUM(II) ION (Mg, NON-CHELATE), substance concentration	0,57	$\text{mmol/l}$
M10	Pt—mass	70,0	kg
M11	U—MELANIN + MELANOGEN, arbitrary substance concentration(method; 0-1)	0	arb. unit

Note 8: An alternative is 'Pt—PATIENT, mass' or 'Pt—BODY, mass'.

Note 8: Note 8: An alternative is 'Pt—PATIENT, mass' or 'Pt—BODY, mass'.

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## RECOMMENDATION 1973

Ser. No.	System	Component(specifications), kind of quantity(specifications)	Number	Unit	Ref.
M12	U—METHYLKETONES, arbitrary substance concentration(Acetest®; 0-1)	0	arb. unit	$10^9/l$	
M13	B—MONOCYTES, number concentration	0,50	arb. unit	$10^9/l$	
M14	(B)Leukocytes—MONOCYTES, number fraction	0,05	1		
M15	S—MUCOPROTEIN, mass concentration(method)	0,75	g/l		
M16	Expectorate— <i>Mycobacterium</i> sp., arbitrary number concentration(method; 0-1)	0	arb. unit	$10^9/l$	
N01	Urethral secretion— <i>Neisseria</i> sp., arbitrary number concentration(method; 0-1)	0	arb. unit	$10^9/l$	
N02	B—NEUTROPHILOCYTES(NON-SEGMENTED), number concentration	0,20	1		
N03	(B)Leukocytes—NEUTROPHILOCYTES(NON-SEGMENTED), number fraction	0,02	1		
N04	B—NEUTROPHILOCYTES(SEGMENTED), number concentration	6,5	$10^9/l$		
N05	(B)Leukocytes—NEUTROPHILOCYTES(SEGMENTED), number fraction	0,65	1		
N06	dF—NITROGEN, amount of substance	70	mmol		
N07	dU—NITROGEN, amount of substance	1,10	mol		
N08	(PPt)S—NITROGEN(AMINO ACID), substance concentration	3,0	mmol/l		
N09	S—NITROGEN(NON-PROTEIN), substance concentration	19	mmol/l		
001	S—ORNITHINE CARBAMOYL TRANSFERASE, catalytic concentration(method)	? ?	kat/l		
002	S— $\alpha$ -OROSOMUCOID, arbitrary substance concentration(method; 0-3)	2	arb. unit	$10^9/l$	
003	(B)Erythrocytes—OSMOTIC PRESSURE REACTION, arbitrary pressure(method)	140	arb. unit	$10^9/l$	
004	dU—OXALATE, amount of substance	0,30	mmol		
005	dU—17-OXOSTEROID, amount of substance(method)	44	$\mu\text{mol}$		
006	(aB)Hemoglobin—OXYHEMOGLLOBIN, mole fraction(method)	0,95	1		
P01	U—PHENYL PYRUVATE, arbitrary substance concentration(method; 0-1)	0	arb. unit	$10^9/l$	
P02	dU—PHOSPHATE(P), amount of substance	35	mmol		
P03	dU,U—PHOSPHATE(P), substance concentration	29	mmol/l		
P04	S—PHOSPHATE(P, NON-ESTERIFIED), substance concentration	1,3	mmol/l		
P05	S—PHOSPHOLIPID(P), substance concentration	2,75	mmol/l		
P06	aB—PLASMA, pH(37,0 °C)	7,41	1		H15
P07	Pt—PLASMA, volume	2,50	1		
P08	B—PLATTULOCYTES, number concentration	225	$10^9/l$		
P09	U—PORPHOBILINOGEN, arbitrary substance concentration(method; 0-1)	0	arb. unit	$10^9/l$	
P10	U—PORPHYRINS, arbitrary substance concentration(method)	? ?	arb. unit	$10^9/l$	
P11	dU—POTASSIUM ION, amount of substance	69	mmol		
P12	dU,U—POTASSIUM ION, substance concentration	58	mmol/l		
P13	Erythrocytes—POTASSIUM ION, substance concentration	90	mmol/l		

LIST OF QUANTITIES IN CLINICAL CHEMISTRY

P14	P <sub>S</sub> —POTASSIUM ION, substance concentration	4,3 mmol/l
P15	dU—PREGNANDIOL, amount of substance	10 μmol
P16	dU—PREGNANTRIOL, amount of substance	12 μmol
P17	U—PROTEIN, arbitrary substance concentration(Albustix® ; 0-1)	0 arb. unit
P18	U—PROTEIN, arbitrary substance concentration(Bence Jones 1948; 0-1)	0 arb. unit
P19	dU—PROTEIN, mass(method)	60 mg
P20	S—PROTEIN, mass concentration(method)	71 g/l
P21	Sf—PROTEIN, mass concentration(method)	0,30 g/l
P22	U—PROTEIN, mass concentration(method)	0,05 g/l
P23	Pt—PULSE, frequency(30 s)	1,20 Hz
R01	Pt—RECTUM, Celsius temperature	36,9 °C
R02	Pt—RECTUM, thermodynamic temperature	31,0,1 K
R03	Pt—RESPIRATION, frequency(60 s)	0,20 Hz
R04	B—RETICULOCYTES, number concentration	10 <sup>9</sup> /l
R05	(B)Erythrocytes—RETICULOCYTES, number fraction	10 <sup>-3</sup>
S01	U—SACCHARIDE, arbitrary substance concentration(Fehling; 0-1)	6 arb. unit
S02	B—SEDIMENTATION REACTION, arbitrary length(method)	0 arb. unit
S03	dU—SODIUM ION, amount of substance	5 mmol
S04	dU,U—SODIUM ION, substance concentration	200 mmol/l
S05	P <sub>S</sub> ,S—SODIUM ION, substance concentration	167 mmol/l
S06	P—SOMATOTROPINE, substance concentration	141 nmol/l
S07	(B)Hemoglobin—SULPHHEMOGLOBIN, mole fraction	0,11 1
S08	Pt—SULPHOBROMOPHTHALENATE, relative amount of substance (Pt 2,7 ks/Pt 0 ks; method)	0,01 0,03
T01	dU—TESTOSTERONE, amount of substance(male Pt)	1 μmol
T02	S—TESTOSTERONE, substance concentration(male Pt)	0,25 nmol/l
T03	(Pt)S—THYMOL REACTION, arbitrary substance concentration(Maclagan 1944; 0, 1, 2...)	0 arb. unit
T04	S—THYROXIN, substance concentration(method)	102 nmol/l
T05	S—β <sub>1</sub> -TRANSFERRIN, arbitrary substance concentration(male Pt)	2 arb. unit
T06	(Pt)S—TRANSFERRIN(74 000), substance concentration	30,8 μmol/l
T07	(Pt)S—TRIACYL GLYCEROL LIPASE(ATOXYL® RESISTANT), catalytic concentration(method)	? kat/l
T08	S—TRIGLYCERIDE, substance concentration(method)	? mmol/l
T09	(Pt)S—TRIODOTHYRONINE REACTION, arbitrary substance concentration(method)	0,75 arb. unit
T10	F—TRYPsin, arbitrary catalytic concentration(method; 0-1)	? 0
T11		

Note 9: EC 3.1.1.3; the component (without specification) was formerly called lipase.

## RECOMMENDATION 1973

Ser. No.	System—Component(specifications), kind of quantity(specifications)	Number × Unit	Ref.
U01	dU—URATE, amount of substance	2,90	mmol
U02	(U)Concrent—URATE, arbitrary substance content(murexide; 0-1)	1	arb. unit
U03	dU,U—URATE, arbitrary substance content(method)	2,42	mmol/l
U04	S—URATE, substance concentration(method)	0,28	mmol/l
U05	P <sub>1</sub> —URINE, mass density(20.0 °C)	1,019	kg/l
U06	P <sub>1</sub> —URINE, pH(indicator paper)	6,3	1
U07	P <sub>1</sub> —URINE, relative density(U 20.0 °C/Water 20.0 °C)	1,020	1
U08	dP <sub>1</sub> —URINE, volume	1,20	1
U09	U—UROBILIN, arbitrary substance concentration(method)	10	arb. unit
U10	(P <sub>1</sub> )U—UROBILINOGEN, arbitrary substance concentration(method; 0-1)	0	arb. unit
U11	dU—UROPORPHYRINS (I + III), amount of substance	10	nmol

LIST OF QUANTITIES IN CLINICAL CHEMISTRY

**10. INDEX**

Acetone bodies	M12	Ehrlich's reaction	≡U10
Acid fast bacilli	M16	Enzyme, cf. IUPAC-IUB (1972)	
A/G ratio	≡A13	Recommended Name	
ALA	A09, A22, A23	Eosinophil count	E01, E02
Albumen, cf. protein		Erythrocyte count	≡E09
Albumin (per cent)	A10, A11, A14, A15	Erythrocyte sedimentation rate	≡S02
Albumin per cent	A12, A13	ESR	≡S02
Aldolase	F09	Excretion, cf. component name	
Alkali reserve	≡C18		
Alkali-stable hemoglobin	≡F05		
Alkapton	≡H13	Faecal, cf. component name	
Alpha-component name cf. α-Component name		Fasting, cf. component name	
Alpha-hydroxybutyric dehydrogenase	H17	Fat, cf. lipid in this Index	
Amino acid, analogously to	H22	Fatty acids	A17, A18, A19
Ammonia	A25, A26	FFA	A17
B <sub>12</sub>	C37	FIGLU	F07
Basal metabolic rate	D05	Folic acid	F06
Base excess	≡B05	Free fatty acids	A17
BB	≡B04		
BE	≡B05	Gerhardt's reaction	M12
Bence Jones protein	P18	GFR	F03
Benzidine reaction	B14, H06, H07	Glomerular filtration rate	F03
Bicarbonate	H14	Glutamic-alanine transaminase	A09
Bleeding time	≡C09	Glutamic-aspartic transaminase	A34
Blood, cf. component name		Glutamic-oxaloacetic transaminase	A34
Blood content	≡B16	Glutamic-pyruvic transaminase	A09
Blood per cent	H11	Gmelin's reaction	B12
Blood pressure	B17, B18	Gonococci	N01
Blood urea	C12	GOT	A34
BMR	D05	GPT	A09
Body temperature	R01, R02		
Bromsulphthalein	S08	Haematocrit	≡E11
BSP	S08	Halometry	≡E04
Buffer base	≡B04	HBD(H)	H17
BUN	C12	Hemoglobin F	≡F05
Catecholamines	A07, A08	5-HIAA	H20
Chlorine, cf. chloride		Hormone, cf. chemical name	
Citric acid	C33		
CO, cf. carbon monoxide		Ig, cf. immunoglobulin	
Colorimetric index	≡H08		
Concentration, cf. component name		Ketone bodies	M12
CPK	C46	Ketosteroid	≡O05
Creatinine, cf. creatininum			
Creatinine clearance	F03	LAP	A24
Delta-component name, cf. δ-Component name		LDH	L02
Density	U05, U07	Legal's reaction	M12
Differential count, cf. component name		Leucine aminopeptidase	A24
Diuresis	U08	Leucocyte count	≡L07
		Lipase	L08
		Lipid	T07, T08
			L10, L11, L12, A18, A19

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MCD	≡E04	Reticulocyte count	R04, R05
MCH	≡H08		
MCHC	≡H09		
MCV	≡E05	Saturation index	≡H09
Mean cell, cf. MC in this Index		Sediment count, cf. component name	
Metabolic rate	D05	Sedimentation reaction	≡S02
Methaemoglobin	H05	Serum, cf. component name	
		SGOT	A34
		SGPT	A09
NEFA	A17	Specific gravity	U07, U05
Net acid	≡A04	Specific weight	U05, U07
Non protein nitrogen	≡N09	Standard bicarbonate	≡H14
NPN	≡N09	Sugar, cf. chemical component name	
 Osmolality	F08	T <sub>3</sub>	T10
Osmolarity	107	T <sub>4</sub>	T04
Osmotic fragility test	≡O03	TB	M16
Osmotic resistance	≡O03	Thrombocyte count	≡P08
Oxygen, cf. dioxygen		Thymol turbidity	≡T03
Oxygen saturation	≡O06, D03, D04	TIBC	T06
		Total, cf. component name	
		Total carbon dioxide	≡C18
Packed cell volume	≡E11	Total iron binding capacity	T06
PBI	I06	TWBC	≡L07
pCO <sub>2</sub>	≡C13		
	C14		
PCV	≡E11	Urea, cf. carbamide	
pH	P06, U06	Uric acid, cf. urate	
Phosphorus, cf. phosphate		Urinary, cf. component name	
Plasma, cf. component name			
Platelet count	≡P08		
PO <sub>2</sub>	≡D03	Vanillylmandelic acid	H21
	D04	Vitamin, cf. chemical name	
P-P	≡C36	Vitamin B <sub>12</sub>	C37
Protein bound iodine	I06	Vitamin C	A33
Prothrombin	C35, C36	VMA	H21
		Volumetric index	≡E11
 Quick time	≡C35		
		Weight, cf. mass	
Red cell count	≡E09	White cell count	≡L07
			L08