Molecular rearrangements in derivatives of grandiflorenic acid [(-)-kaur-9(11),16-dien-19-oic acid]

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Abstract: Seven new skeletal diterpenes, 4,6,8,9,11,13 and 14, were obtained *via* the rearrangements of 1b,c, 3, 10a,b,c and 12, the derivatives of 1a, and the structures and stereochemistries of these compounds and also the mechanisms for their formation are described.

The objective of this work is to study the rearrangements in the derivatives of grandiflorenic acid 1a [(-)-kaur-9(11),16-dien-19-oic acid] in order to construct diterpenes with new ring systems. Our initial investigation aimed to transform 1a to the analogs of zoapatlin 2a (1), eupatalbin 2b, or eupatoralbin 2c (2). We first investigated this possibility via the rearrangements of the 9,11-epoxide of methyl (-)-kaur-9(11)-en-19-oate 1b. Epoxidation of 1b with MCPBA took place stereoselectively at the more accessible α -side and yielded exclusively 9, 11α -epoxide 3. During this epoxidation we observed that when 1b contained N-nitrosomethylurea as an impurity, the expected 3 was not obtained, but instead 4 was isolated.

1a; R1=H, R2=CH2

b; $R^1=CH_3$, $R^2=\beta-CH_3$, $\alpha-H$

c; R1=CH3, R2=O

$$\bigcap_{C=0}^{R^1} \mathbb{R}^3$$

2a; R1=CH₂, R2=R3=H₂

b; $R^1 = \beta - CH_3, \alpha - OH, R^2 = \alpha - OH, \beta - H, R^3 = H_2$

c; $R^1=\beta$ -CH₃, α -OH, $R^2=H^2$, $R^3=\alpha$ -OH, β -H

$$HO$$
 HO
 HO
 HO
 CO_2CH_3
 CO_2CH_3

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We studied in detail how this reaction proceeded. As a consequence it was found that 3 was an intermediate in this reaction pathway and nitrosomethylurea might generate in the presence of MCPBA a catalytic amount of HNO₃ which would act as a Lewis acid to cleave the epoxide ring to form 6a. 6a would then react with one mol equiv of MCPBA to give 7, which upon cleavage of the epoxide ring would afford 4, as illustrated in Scheme 1.

Scheme 1

That 6a and 7 were indeed involved in this reaction pathway was demonstrated by an independent experiment in which when 6a, obtained via a different route (see below), was epoxidized and the resulting 7 was treated with a catalytic amount of HCl, 4 was formed. On treatment with BF₃-Et₂O in benzene 3 afforded 6a. Assuming that in this case the intermediate alcohol 5, if formed, might be acetylated in situ in the presence of Ac₂O to produce 8a* rather than 6a, we treated 3 with BF₃-Et₂O in Ac₂O. What we obtained, however, was not the expected 8a but 6b, which we presume would be generated by the mechanism shown in Scheme 2.

4 and 6 represent new diterpene skeletons since so far only diterpenes of the pimarane-, cassaneand cleisthanthane-type are known to occur in nature.

MacMillan, et al. (3) reported that on treatment with acetyl hypobromite or Br₂ 1c underwent $10\alpha \rightarrow 9\alpha$ -methyl migration to yield 11β -bromo derivative 9a. This indicated that in this case bromination apparently occurred from the β -side of the molecule. The lack of the sterically shielding 16β -methyl group in 1b would allow a bromine molecule to attack from the β -side. The β -stereochemistry of the intermediate brominium ion would greatly facilitate subsequent migration of the α -oriented 20-methyl group to C-9. 9b was synthesized from 9a by dehydrobromination, followed by hydrogenation of the resulting 11,12-double bond and subsequent Wittig reaction with methylenetriphenylphosphorane.

^{*}Note that this compound is a ring C/D-analog of antheridiogen, A_{AN}, the antheridium-inducing factor from *Anemia phyllitidis* (see ref.4)

$$\begin{array}{c} \textbf{1c} \\ \textbf{9a;} \ R^1 = \beta \text{-Br}, \alpha \text{-H}, \ R^2 = 0 \\ \textbf{b;} \ R^1 = H_2, \ R^2 = \text{CH}_2 \\ \textbf{c;} \ R^1 = 0, \ R^2 = \beta \text{-CH}_3, \alpha \text{-H} \end{array}$$

On treatment with a catalytic amount of H_2SO_4 10a rearranged to give 8a (96%). On the other hand, the rearrangement of 10b under identical conditions afforded 6a (13%) and 8b (8%). Our next interest was directed to a pinacolic type rearrangement of 10b. Treatment of 10c with potassium t-butoxide in t-butanol provided 11^* (100%).

On treatment with BF₃ - Et₂O in Ac₂O-AcOH 12 did not undergo a pinacolic type rearrangement, but it yielded instead 13 (14%) and 9c (33%). On the other hand, on treatment with BF₃ - Et₂O in Ac₂O 12 suffered a profound backbone rearrangement to form a new ring B aromatic diterpene 14 (33%). A plausible mechanistic pathway for its formation is depicted in Scheme 3.

An independent experiment, in which 9c, obtained via a different route (see above), afforded 14 on treatment with BF3 - Et2O in Ac2O, verified that 9c was indeed an intermediate in this reaction pathway. In Scheme 3 it is shown how each carbon migrates during this rearrangement process by numbering all the carbon atoms in 12. If C-12 in 12 is labeled by deuterium, then the deuterium must be detected in the same numbered carbon atom in 14, provided that our postulated mechanistic pathway is correct. On treatment with CD₃OD, D₂O and sodium 12 afforded a mixture of 12-d₀-12 (66%), 12-d₁-12 (27%) and 12,12-d₂-12 (6%). In its broad band proton decoupled ¹³C n.m.r. spectrum, besides all the carbon resonances in do - 12, four other carbon signals appeared [8 44.81 (br t), C-12; δ 40.39 (s), C-13; δ 34.99 (s), C-16] whose signal positions are slightly upfield relative to the corresponding carbons in d₀ - 12, due to the isotope effect (6). The carbonyl C-11 experienced a downfield shift (7) to 8 215.87 (s). On treatment with BF3 - Et2O in Ac2O under identical conditions, the above mixture of compounds yielded a mixture of do - 14 and its d1-derivative. Its broad band proton decoupled ¹³C n.m.r. spectrum was identical with that of d₀-14 except that in this case three additional carbon signals were observed. A deuterated carbon was found at 8 30. 57 (br t), which was upfield by 0.29 ppm, as compared with the corresponding C-12 in do-14. Its neighboring C-13 and -16 resonated at δ 39.02 (s) and δ 36.50 (s), respectively, exhibiting an expected upfield

^{*}The toxic substances, grayanotoxins I, II and III are among ring B-homo diterpene analogs with a rearranged kaurane skeleton (see reference 5).

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isotope shift. All these n. m. r. results attested to the correctness of the mechanistic pathway which we depicted in Scheme 3 for the formation of 14 from 12. The structures and stereochemistries of all new compounds obtained were assigned on the basis of spectroscopic data and further confirmed by X-ray crystallographic analysis (4, 6b, 8a, 13 and 14).

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