# Synthesis of oxygen heterocycles via alkynyltungsten compounds\*

Rai-Shung Liu\*

Department of Chemistry, National Tsing-Hua University, Hsinchu, Taiwan ROC

Abstract: This short review article covers some useful applications of alkynyltungsten compounds to the syntheses of complex lactones. Two types of cyclizations will be emphasized: (1) cycloalkenylation of tungsten-alkynol compounds with aldehydes to give  $\alpha$ -alkylidene oxacarbeniums, further leading to  $\alpha$ -alkylidene lactones and (2) intramolecular [3+2]-cycloaddition of epoxides to give bicyclic lactones. The new methodologies can provide a short synthesis of enantiopure lactones such as (–)-epilitsenolide  $C_2$ , (+)-listenolide  $C_1$ , (+)-isodihydromahubanolide A, (+)-blastmycinone, and (–)-epi-blastmycinone.

### INTRODUCTION

The reaction of alkynylmetal complexes with organic carbonyl compounds can lead to oxygenated molecules; the regioselectivities depend on the types of metals as shown in Scheme 1 [1–2]. Alkynyl organometallics of silanes, boranes, stannanes, and zinc are not as useful as their allyl-, propargyl and allenyl species because of their low reactivities [1]. These alkynyl organometallics undergo electrophilic addition at their  $C_{\alpha}$ -carbons to generate an unstable vinyl cation that is easily captured by any basic species in solution to give alkynyl derivative. Alkynyl compounds of electron-rich transition metals however show a distinct reaction pathway [2–3]. These organometallics react with carbon electrophiles at their  $C_{\beta}$ -carbons to form metal-vinylidenium intermediates which are fairly kinetically stable. Nucleophilic attack at cations of these types proceeds with regiochemistry at their  $C_{\alpha}$ -carbons to effect a 1,2-addition product. Metal-vinylidenium species is also a valuable intermediate in various catalytic carbon–carbon bond formations [4]. In this study, we report two useful cyclizations based on alkynyltungsten compounds, and each reaction involves sequential bond-breaking and -forming processes.

(1) 
$$M = R$$
  $C + M$   $C + R$   $C - R$ 

Scheme 1

## **CYCLOALKENYLATIONS**

Treatment of tungsten alkynols with aldehydes and BF<sub>3</sub>·Et<sub>2</sub>O (1.0 equiv) in diethyl ether produced tungsten-oxacarbenium salts in quantitative yields (>98%) [5]. The structures of tungsten-oxacarbeniums

<sup>\*</sup>Lecture presented at the XIX<sup>th</sup> International Conference on Organometallic Chemistry (XIX ICOMC), Shanghai, China, 23–28 July 2000. Other presentations are published in this issue, pp. 205–376.

266 R.-S. LIU

(eqs. 1–2, Scheme 2) were characterized by X-ray diffraction studies. The oxacarbenium salts were isoelectronic with neutral Group 6 Fisher carbene complexes, and easily demetallated by  $H_2O$ /air to afford  $\alpha$ -alkylidene lactones in good yields [5]. This synthetic method is applicable to other electrophiles such as aliphatic aldehydes and trimethoxymethane to give good yields of  $\alpha$ -alkylidene  $\gamma$ - and  $\delta$ -lactones. This method is also effective for the synthesis of  $\alpha$ -alkylidene  $\gamma$ -lactams derived from alkynyltungsten amines and aldehydes [6]. Attempts to extend this cyclization for the synthesis of larger  $\epsilon$ -lactones and  $\gamma$ -lactams were unsuccessful.

(1) W = OH 
$$\frac{PhCHO}{BF_3Et_2O}$$
  $\frac{H_2O/air}{Ph}$   $\frac{Me}{BF_3OH}$   $\frac{H_2O/air}{Ph}$   $\frac{Me}{BF_3OH}$   $\frac{H_2O/air}{Ph}$   $\frac{H_2O/air}{BF_3OH}$   $\frac{H_2O/air}{Ph}$   $\frac{H_2O/air}{BF_3OH}$   $\frac{H_2O/air}{Ph}$   $\frac{H_2O/air}{BF_3OH}$   $\frac{H_2O/air}{Ph}$   $\frac{H_2$ 

#### Scheme 2

The oxacarbenium salts can undergo cyclopropanation with diazomethane to provide a one-pot synthesis of spiro- $\epsilon$ - and  $\gamma$ -lactones as depicted in Scheme 3. The reaction proceeds with high diastere-oselectivities with a new cyclopropane ring trans to the phenyl or methyl substituent. Notably, there are two distinct pathways. The oxacarbenium salt bearing one alkylidene group uptakes one molecule of  $CH_2N_2$ , whereas those having an arylidene group uptake two molecules. According to isotopic labeling experiment, one of the  $CH_2$  inserts into the Ar–C single bond of the arylidene group. The mechanism is proposed to involve a bridging phenonium ion [5].

$$W = CpW(CO)_{3}$$

$$R' = Me, Ph$$

$$R' = Me, Ph$$

$$R = RHO_{1}$$

$$R' = RHO_{2}$$

$$R' = RHO_{1}$$

$$R' = RHO_{2}$$

$$R' = RHO_{1}$$

$$R' = RHO_{2}$$

$$R'' = RHO_{2}$$

$$R'' = RHO_{3}$$

$$R'' = RHO_{4}$$

$$R'' = RHO_{4}$$

$$R'' = RHO_{4}$$

$$R'' = RHO_{4}$$

#### Scheme 3

Intramolecular cycloalkenation of alkynyltungsten compounds is a useful method for the synthesis of bicyclic unsaturated lactones [7]. The sizes can be up to seven-membered carbocyclic rings fused with  $\gamma$ - and  $\epsilon$ -lactones. The electrophiles include a tethered dimethoxymethane, trimethoxymethanes, and ketone group. One advantage here is that starting alkynyltungsten species bearing a tethered electrophile and alcohol were easily prepared in two steps from the alkynols HCC(CH<sub>2</sub>)<sub>n+1</sub>OH and bromoalkanyl dimethoxymethane. The dimethoxymethane and trimethoxymethane derivatives gave better yields compared to their ketone analogs. An application of this method is a short synthesis of natural mitsugashilactones and onikulactone. Mitsugashilactone was selectively prepared via treatment of tungsten-pentynol (eq. 3) with BF<sub>3</sub>·Et<sub>2</sub>O in diethyl ether, followed by sequential treatment of oxacarbenium intermediates with Me<sub>2</sub>CuLi and water. A similar approach gave onikulactone in reasonable yield with a different tungsten-pentynol complex (eq. 4).

We were successful in extending cycloalkenylations to the synthesis of enantiopure natural lactones [8–10], such as (–)-epilitsenolide  $C_2$ , (+)-listenolide  $C_1$ , (+)-isodihydromahubanolide A, (+)-blast-mycinone and (–)-epi-blastmycinone. Scheme 5 shows the key step for total synthesis of (+)-blastmycinone and (+)-antimycinone based on a modified cycloalkenylation. We employed the dimethoxymethoxy derive (Scheme 5) and alkynyl aldehydes to achieve the reaction, giving the desired

Scheme 4

oxacarbenium salts, further leading to unsaturated lactones in good yields. Aliphatic aldehydes did not work in this case, and tungsten-furayl complexes were found as major products. The overall yield of (+)-blastmycinone and (+)-antimycinone is ca. 23–25% yields in nine steps from (S)-methyl lactate. Natural  $\alpha$ -alkylidene  $\gamma$ -lactones such as (–)-epilitsenolide  $C_2$ , (+)-listenolide  $C_1$ , (+)-isodihydrom-ahubanolide, and trisubstituted lactone (–)-epi-blastmycinone were also successfully prepared from this cyclization.

(S)-methyl lactate 
$$\frac{4\text{-steps}}{48\%}$$
  $\frac{4\text{-steps}}{\text{MOMO}}$   $\frac{(1) \text{ BF}_3\text{:Et}_2\text{O}}{\text{R}}$   $\frac{3\text{-steps}}{\text{OMOM}}$   $\frac{3\text{-steps}}{\text{R}}$   $\frac{3\text{-steps}}{\text{OMOM}}$   $\frac{3\text{-steps}}{\text{R}}$   $\frac{3\text{-steps}}{\text{CHO}}$   $\frac{3\text{-steps}}{\text{R}}$   $\frac{3\text{-steps}}{\text{CHO}}$   $\frac{3\text{-steps}}{\text{R}}$   $\frac{3\text{-steps}}{\text{C}}$   $\frac{3\text{-steps}}{\text{R}}$   $\frac{3\text{-steps}}{\text{R}}$ 

Scheme 5

### **CYCLOADDITION OF EPOXIDES**

Although epoxides and aziridines are important reagents, cycloadditions of epoxides with functionalized alkynes remains unknown. [3+2]-Cycloaddition of functionalized alkenes with aziridine is reported recently [11]. We recently discovered [12] that alkynyltungsten compounds undergo smooth cycloaddition with epoxides via vinylidenium intermediates. Such [3+2]-cycloadditions proceed with high diastereoselectivities, and *trans*-epoxides gave only *cis*-lactones, whereas *cis*-epoxides gave only *trans*-lactones (Scheme 6). This implies a  $S_N 2$  mechanism for ring opening of epoxides. Only *cis*-fused isomers were formed for various bicyclic  $\gamma$ -lactones. In the case of a *trans*-phenyl oxide (eq. 3), the cyclization proceeds with *endo*-attack of epoxide, whereas *cis*-phenyl oxide gave cycloaddition products (eq. 2). The cycloadditions also work well for tethered *trans*-aziridines with a similar fashion. *Cis*-aziridines did not undergo cycloadditions, under conditions, and this may be attributed to steric effects that disfavor an intramolecular  $S_N 2$  process. We also prepared optically active epoxides (eqs. 4–5) to

268 R.-S. LIU

(1) L.A (20 mol%) 
$$R = alkyl;$$
  $R = alkyl;$   $R = alkyl;$ 

#### Scheme 6

study the nature of cyclization. No significant loss of enantiopurities is observed for secondary epoxide (eq. 3), but nearly 24% loss of enantiopuritie is observed for tertiary epoxide (eq. 4).

Scheme 7 shows an instance for cycloaddition of an epoxide that can form a synthetically useful oxacarbenium if the reaction is performed in diethyl ether [13]. In principle, these cationic salts can lead to a number of oxygen heterocycles via treatment with various nucleophiles. Since optically pure epoxides are easily prepared according to literature methods [14]. It is speculated that such cycloadditions will be useful methods for synthesis of natural compounds.

$$W = \bigcap_{GH_{13}} OAc$$

$$W = CpW(CO)_3$$

$$W = OAc$$

$$C_6H_{13} BF_3 OH$$

$$W = CpW(CO)_3$$

$$W = OAc$$

$$C_6H_{13} BF_3 OH$$

$$W = CpW(CO)_3$$

$$W = OAc$$

$$C_6H_{13} BF_3 OH$$

$$C_6H_{13} BF_3 OH$$

$$C_6H_{13} BF_3 OH$$

Scheme 7

## **REFERENCES**

- (a) H. Yamamoto. In Comprehensive Organic Synthesis: Addition to C-X π bonds Part II, B. M. Trost and I. Fleming (Eds.), Vol. 2, Ch. 1.3, p. 81, Pergamon, Oxford (1991); (b) I. Fleming, In Comprehensive Organic Synthesis: Addition to C-X π Bonds, Part I; B. M. Trost and I. Fleming (Eds.), Vol. 2, Ch. 2.2, p. 575, Pergamon Oxford (1991); (c) A. Hosomi. Acc. Chem. Res. 21, 200 (1988); (d) H. Sukurai. Synlett 1 (1989).
- (a) M. Rosenblum. *J. Organometal. Chem.* 300, 191 (1986); (b) M. Rosenblum. *Acc. Chem. Res.* 7, 122 (1974); (c) A. Wojcicki and C. E. Schuchart. *Chem. Rev.* 105, 35 (1990); (d) M. E. Welker *Chem. Rev.* 92, 97 (1992).
- 3. C.-L. Lei and R.-S. Liu. Chem Rev. 100, 3127 (2000).
- 4. C. Bruneau and P. H. Dixneuf. Acc. Chem. Res. 311, 32 (1999).
- 5. K.-W. Liang, W.-T. Li, G.-H. Lee, S.-M. Peng, R.-S. Liu. J. Am. Chem. Soc. 119, 4404 (1997).
- 6. S.-J. Chen, S.-T. Chang, R-S. Liu. Tetrahedron 56, 5029 (2000).
- 7. K.-W. Liang, M. Chandrasekharam, C.-L. Li, R.-S. Liu. J. Org. Chem. 63, 7289 (1998).
- 8. M.-J. Chen, C.-Y. Lo, C.-C Chin, R.-S. Liu. J. Org. Chem. 65, 6362 (2000).

- 9. M.-J. Chen, C.-Y. Lo, R.-S. Liu, Synlett 1205 (2000).
- 10. M.-J. Chen, C.-Y. Lo, C.-C. Chin, R.-S. Liu. Tetrahedron Lett. (2001). In press.
- 11. N. Masako and K. Michiaki. Org. Lett. 2, 953 (2000).
- 12. T.-L. Chen, R.-J. Madhushaw, R.-S. Liu. Submitted for publication.
- 13. T.-L. Chen and R.-S. Liu. Unpublished results.
- 14. Z.-X. Wang, Y. Tu, M. Frohn, J. R. Zhang, Y. Shi. J. Am. Chem. Soc. 119, 11224 (1997).